Johnson '204 and '852 author search/

Page 1

=> d 150 1-5 all

synthesis

Aluminophosphate zeolites

IT

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L50
     ANSWER 1 OF 9 HCAPLUS COPYRIGHT 2004 ACS on STN
AN
     2004:602764 HCAPLUS
DN
     141:129581
     Entered STN: 28 Jul 2004
ED
     Synthesis of alumino- and silicoalumino-phosphates of CHA framework
TΙ
IN
     Cao, Guang; Shah, Matu J.
PΑ
     Exxonmobil Chemical Patents Inc., USA
SO
     U.S., 19 pp.
     CODEN: USXXAM
DT
     Patent
LA
     English
IC
     ICM B01J027-182
     ICS C01B039-54
     502214000; 502208000; 423328100; 423328200; 423329100
NCL
     67-1 (Catalysis, Reaction Kinetics, and Inorganic Reaction
CC
     Mechanisms)
     Section cross-reference(s): 51
FAN.CNT 1
     PATENT NO.
                                      APPLICATION NO.
                         KIND
                                DATE
                                                                   DATE
PΤ
     US 6767858
                         B1
                                20040727 US 2003-370932
                                                                   200302
                                                                   20
PRAI US 2003-370932
                                20030220
CLASS
 PATENT NO.
                 CLASS PATENT FAMILY CLASSIFICATION CODES
 US 6767858
                 ICM
                        B01J027-182
                 ICS
                        C01B039-54
                        502214000; 502208000; 423328100; 423328200;
                 NCL
                        423329100
    The invention is directed to a method of synthesizing
AB
    aluminophosphate and silicoaluminophosphate mol. sieves and in
     particular to the synthesis of aluminophosphate and
    silicoaluminophosphate mol. sieves using N-methylethanolamine as
    template with or without a source of fluoride. The use of
    N-methylethanolamine as template results in good quality AlPO4 of
    CHA framework type and SAPO mol. sieves of CHA framework type with
     low levels of silicon in high yield.
    aluminophosphate silicoaluminophosphate zeolite CHA framework type
ST
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(AlPO4; synthesis of alumino- and silicoalumino-phosphates of CHA

framework type)

IT Silicoaluminophosphate zeolites

(SAPO; synthesis of alumino- and silicoalumino-phosphates of CHA framework type)

IT Catalysts

(synthesis of alumino- and silicoalumino-phosphates of CHA framework type)

IT Amines, processes

Quaternary ammonium compounds, processes

(synthesis of alumino- and silicoalumino-phosphates of CHA framework type)

IT 7784-30-7P, Aluminum phosphate

(synthesis of alumino- and silicoalumino-phosphates of CHA framework type)

ΙT 56-34-8, Tetraethylammonium chloride 71-91-0, Tetraethylammonium 75-31-0, Isopropylamine, processes 77-98-5, Tetraethylammonium hydroxide 108-91-8, Cyclohexylamine, processes 109-83-1, N-Methylethanolamine 110-68-9, Methylbutylamine 110-91-8, Morpholine, processes 111-42-2, Diethanolamine, 121-44-8, Triethylamine, processes processes 142-84-7, Dipropylamine 429-07-2, Tetraethylammonium hexafluorophosphate 665-46-3, Tetraethylammonium fluoride 1185-59-7, Tetraethylammonium acetate 1333-83-1, Sodium bifluoride (nahf2) 1341-49-7, Ammonium hydrogen difluoride 7664-39-3, Hydrogen 7784-18-1, Aluminum trifluoride fluoride, processes Triammonium hexafluoroaluminate 16919-19-0, Ammonium 16919-24-7, Diammonium hexafluorostannate hexafluorosilicate 16919-31-6, Diammonium hexafluorozirconate 16940-81-1, Hydrogen 16961-83-4, Dihydrogen hexafluorosilicate hexafluorophosphate 16962-40-6, Diammonium hexafluorotitanate 16962-47-3, Diammonium hexafluorogermanate 16984-48-8, Fluoride, processes 21324-39-0, Sodium hexafluorophosphate 32287-65-3, Aluminum trifluoride 71494-19-4, Tetraethylammonium phosphate hydrate (synthesis of alumino- and silicoalumino-phosphates of CHA

(synthesis of alumino- and silicoalumino-phosphates of CHA framework type)

RE.CNT 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Anon; EP 0993867 2000 HCAPLUS
- (2) Guth; US 5096684 A 1992 HCAPLUS
- (3) Kuehl; US 4786487 A 1988 HCAPLUS
- (4) Lillerud; US 6001328 A 1999 HCAPLUS
- (5) Liu; US 6162415 A 2000 HCAPLUS
- (6) Lok; US 4440871 A 1984 HCAPLUS
- (7) Pellet; US 4861739 A 1989 HCAPLUS
- (8) Wilson; US 4310440 A 1982 HCAPLUS

L50 ANSWER 2 OF 9 HCAPLUS COPYRIGHT 2004 ACS on STN AN 2003:991131 HCAPLUS

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DN
     140:29334
ED
     Entered STN:
                   21 Dec 2003
     Dimethylamino group-containing amines and alkanolamines as templates
TI
     for synthesis of chabazite-type SAPO zeolite catalysts, especially
     for methanol conversion to lower alkenes
IN
     Cao, Guang; Shah, Matu J.
PA
     Exxonmobil Chemical Patents, Inc., USA
SO
     U.S. Pat. Appl. Publ., 17 pp.
     CODEN: USXXCO
DT
     Patent
LA
     English
IC
     ICM B01J027-182
NCL
     502214000
CC
     51-11 (Fossil Fuels, Derivatives, and Related Products)
     Section cross-reference(s): 49
FAN.CNT 4
     PATENT NO.
                         KIND
                                             APPLICATION NO.
                                                                     DATE
                                 DATE
PΙ
     US 2003232718
                          Α1
                                 20031218
                                             US 2002-170293
                                                                     200206
                                                                     12
     US 6680278
                          B2
                                 20040120
     US 2003232006
                          A1
                                 20031218
                                             US 2003-422923
                                                                     200304
                                                                     24
     WO 2003106343
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                                 20031224
                                             WO 2003-US12713
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             LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ,
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             GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
    WO 2003106341
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                                          WO 2003-US18358
                                                                     200306
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             NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ,
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TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM,
             AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
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PRAI US 2002-170293
                          A2
                                 20020612
     US 2002-171186
                          Α
                                 20020612
     US 2002-171257
                          Α
                                 20020612
     US 2003-422923
                          Α
                                 20030424
CLASS
 PATENT NO.
                 CLASS
                         PATENT FAMILY CLASSIFICATION CODES
 US 2003232718
                 ICM
                         B01J027-182
                 NCL
                         502214000
OS
     MARPAT 140:29334
AΒ
     Chabazite-type silicoaluminophosphate (SAPO)-type zeolites are
     synthesized by reaction of alumina, silica, and phosphorus source
     materials in the presence of an amine and alkanolamine template
     contg. a dimethylamino group and (optionally) a tetraethylammonium
     quaternary ammonium compd. The product SAPO zeolites have
     characteristic x-ray powder diffraction patterns peaks
     (2\theta:d\text{-spacing (in Å}):\text{relative intensity) of: (1)}
     20.62:4.30:100; (2) 9.461:9.34:65; (3) 16.018:5.53:50; (4)
     24.74:3.59:47; (5) 30.66:2.91:45; (6) 30.939:2.89:23; (7)
     17.619:5.03:22; and (8) 25.96:3.43:20. Preferred templates are
     RN(CH3)2 (R = aliph. and cycloaliph.), esp. N, N-dimethyl-C2-7-
     alkanolamines, N, N-dimethyl-C2-7-alkylenediamines, and
     N, N-dimethyl-C2-7-alkylamines. The SAPO zeolites have special
     activity in the conversion of oxygenates, esp. MeOH, to light
     olefins (esp. ethylene and propylene).
ST
     SAPO zeolite dimethylamino amine alkanolamine template;
     silicoaluminophosphate zeolite methanol conversion alkene; chabazite
     SAPO zeolite methanol conversion alkene
IT
     Alkenes, preparation
        (C2-4, methanol conversion to; dimethylamino group-contg. amines
        and alkanolamines as templates for synthesis of chabazite-type
        SAPO zeolite catalysts, esp. for methanol conversion to lower
        alkenes)
IT
     Amines, uses
        (alicyclic, templates; dimethylamino group-contg. amines and
        alkanolamines as templates for synthesis of chabazite-type SAPO
        zeolite catalysts, esp. for methanol conversion to lower alkenes)
ΙT
     Alicyclic compounds
        (amines, templates; dimethylamino group-contg. amines and
        alkanolamines as templates for synthesis of chabazite-type SAPO
        zeolite catalysts, esp. for methanol conversion to lower alkenes)
ΙΤ
    Alcohols, uses
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(amino, templates; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙΤ Silicoaluminophosphate zeolites (chabazite-type; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙT Petroleum refining catalysts (conversion; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙΤ Amines, uses (dimethylalkylenediamines, templates; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙT Functional groups (dimethylamino, templates; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) IΤ Chabazite-type zeolites (silicoaluminophosphates; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙΤ Amines, uses (templates; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙΤ Quaternary ammonium compounds, uses (tetraethylammonium salts; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙΤ 67-56-1, Methanol, processes (conversion of; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes) ΙΤ 56-34-8, Tetraethylammonium chloride 66-40-0D, Tetraethylammonium, salts 71-91-0, Tetraethylammonium bromide 77-98-5, Tetraethylammonium hydroxide 108-00-9, N, N-Dimethylethylenediamine 108-01-0, N, N-Dimethylethanolamine 108-16-7, 1-Dimethylamino-2-109-55-7, N, N-Dimethylpropylenediamine 598-56-1, propanol N, N-Dimethylethylamine 665-46-3, Tetraethylammonium fluoride 926-63-6, N, N-Dimethylpropylamine 1938-58-5 3179-63-3, N, N-Dimethylpropanolamine 3529-10-0 4385-04-0, N, N-Dimethylhexylamine 5277-11-2, N, N-Dimethylheptylamine

26153-88-8, N,N-Dimethylpentylamine 154976-19-9,

154976-21-3, Hexanol, (dimethylamino) -

22078-09-7

Butanol, (dimethylamino) -

597578-32-0, Heptanol, (dimethylamino)-

THE STATE OF THE

(templates; dimethylamino group-contg. amines and alkanolamines as templates for synthesis of chabazite-type SAPO zeolite catalysts, esp. for methanol conversion to lower alkenes)

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L50
     ANSWER 3 OF 9 HCAPLUS
                              COPYRIGHT 2004 ACS on STN
     2003:991024 HCAPLUS
ΑN
     140:18202
DN
     Entered STN:
ED
                   21 Dec 2003
ΤI
     Synthesis of CHA framework type-zeolites as petroleum refining
IN
     Cao, Guang; Shah, Matu J.; Strohmaier, Karl G.;
     Hall, Richard B.
PA
     USA
SO
     U.S. Pat. Appl. Publ., 16 pp.
     CODEN: USXXCO
DT
     Patent
LA
     English
IC
     ICM
         C01B037-04
     ICS
          C01B037-08; B01J029-03
NCL
     423305000; 423306000; 502208000; 502214000
     51-11 (Fossil Fuels, Derivatives, and Related Products)
CC
     Section cross-reference(s): 49, 67
FAN.CNT 4
     PATENT NO.
                         KIND
                                 DATE
                                             APPLICATION NO.
                                                                     DATE
     US 2003231999
PΙ
                          A1
                                 20031218
                                             US 2002-171257
                                                                     200206
                                                                     12
     WO 2003106340
                          Α1
                                 20031224
                                             WO 2003-US12718
                                                                     200304
                                                                     24
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             GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
    WO 2003106341
                                             WO 2003-US18358
                          A1
                                 20031224
                                                                     200306
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             GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
PRAI US 2002-170293
                          Α
                                 20020612
     US 2002-171186
                           Α
                                 20020612
     US 2002-171257
                          Α
                                 20020612
     US 2003-422923
                          Α
                                 20030424
CLASS
 PATENT NO.
                 CLASS
                         PATENT FAMILY CLASSIFICATION CODES
 US 2003231999
                  ICM
                         C01B037-04
                 ICS
                         C01B037-08; B01J029-03
                 NCL
                         423305000; 423306000; 502208000; 502214000
     Microporous aluminophosphate or silicoaluminophosphate mol. sieves
AΒ
     having the CHA framework type are prepd. by (a) reacting a source of
     aluminum, a source of phosphorus, optionally a source of silicon, at
     least one source of fluoride ions and at least one template contg.
     one or more N, N-dimethylamino moieties, (b) inducing crystn. of
     aluminophosphate and/or silicoaluminophosphate mol. sieve from the
     reaction mixt.; and (c) recovering and calcining of the mol. sieves.
     The template can be N, N-dimethylbutanolamine, N, N-
     dimethylheptanolamine, N, N-dimethylhexanolamine,
     N, N-dimethylethylenediamine, N, N-dimethylpropylenediamine,
     N, N-dimethylbutylenediamine, N, N-dimethylheptylenediamine,
     N, N-dimethylhexylenediamine, dimethylethylamine,
     dimethylpropylamine, dimethylheptylamine, dimethylhexylamine, or
     preferably N, N-dimethylethanolamine. The source of fluoride can be
     a metal fluoride, ammonium fluoride, tetralkylammonium fluorides, or
     preferably hydrogen fluoride. The fluoride is within the
     intra-cryst. structure of the mol. sieve. The prepd. zeolites can
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ST silicoaluminophosphate aluminophosphate zeolite prepn template fluoride petroleum refining catalyst

be used as catalysts, e.g. for the conversion of methanol to

IT Petroleum refining catalysts

(synthesis of CHA framework type-zeolites as petroleum refining catalysts)

IT Aluminophosphate zeolites

Silicoaluminophosphate zeolites

(synthesis of CHA framework type-zeolites as petroleum refining catalysts)

IT 24623-77-6, Catapal

olefins.

IT 7631-86-9, Cabosil, reactions

(colloidal, Cabosil; synthesis of CHA framework type-zeolites as petroleum refining catalysts)

IT 7664-38-2, Phosphoric acid, reactions

(synthesis of CHA framework type-zeolites as petroleum refining catalysts)

IT 108-00-9, N,N-Dimethylethylenediamine 108-01-0,

N,N-Dimethylethanolamine 109-55-7, N,N-Dimethylpropylenediamine 598-56-1 926-63-6, Dimethylpropylamine 1862-07-3, 1-Hexanol, 6-(dimethylamino) - 1938-58-5, 1,6-Hexanediamine, N,N-dimethyl 3529-10-0, 1,4-Butanediamine, N,N-dimethyl 4385-04-0 5277-11-2,

Dimethylheptylamine 7664-39-3, Hydrofluoric acid, uses 13330-96-6, 1-Butanol, 4-(dimethylamino) - 22078-09-7,

1,7-Heptanediamine, N,N-dimethyl 27384-67-4, 1-Heptanol,

7-(dimethylamino)-

(template; synthesis of CHA framework type-zeolites as petroleum refining catalysts)

L50 ANSWER 4 OF 9 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2003:989895 HCAPLUS

DN 140:18201

ED Entered STN: 19 Dec 2003

TI Synthesis of CHA framework type-zeolites as petroleum refining catalysts

IN Cao, Guang; Shah, Matu J.

PA USA

SO U.S. Pat. Appl. Publ., 17 pp., Cont.-in-part of U.S. Ser. No. 170,293.

CODEN: USXXCO

DT Patent

LA English

IC ICM C01F007-00

NCL 423705000

CC 51-11 (Fossil Fuels, Derivatives, and Related Products) Section cross-reference(s): 49

FAN.CNT 4

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2003232006	A1	20031218	US 2003-422923	200304 24 200206
	US 2003232718	A1	20031218	US 2002-170293	
	US 6680278	В2	20040120		12

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WO 2003106341
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                                 20031224
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             GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
PRAI US 2002-170293
                          A2
                                 20020612
     US 2002-171186
                                 20020612
                          Α
     US 2002-171257
                          Α
                                 20020612
     US 2003-422923
                          Α
                                 20030424
CLASS
 PATENT NO.
                 CLASS
                        PATENT FAMILY CLASSIFICATION CODES
 US 2003232006
                 ICM
                        C01F007-00
                 NCL
                        423705000
AB
     Cryst. silicoaluminophosphate mol. sieves having the CHA framework
     type are prepd. by reacting a source of aluminum, a source of
     phosphorus, a source of silicon and at least one org. template
     contg. one or more N,N-dimethylamino moieties, inducing crystn. of
     the mol. sieve from the reaction mixt., and calcining.
                                                              The template
     can be N, N-dimethylethanolamine, N, N-dimethylpropanolamine,
     N, N-dimethylbutanolamine, N, N-dimethylheptanolamine,
     N, N-dimethylhexanolamine, N, N-dimethylethylenediamine,
     N, N-dimethylbutylenediamine, N, N-dimethylheptylenediamine,
    N, N-dimethylhexylenediamine, 1-dimethylamino-2-propanol,
    N, N-dimethylethylamine, N, N-dimethylpropylamine,
    N, N-dimethylpentylamine, N, N-dimethylheptylamine, and
    N, N-dimethylhexylamine. The org. template can be tetraethylammonium
    chloride, tetraethylammonium bromide, or tetraethylammonium
     fluoride. The prepd. zeolites can be used as catalysts, e.g. for
     the conversion of methanol to olefins.
ST
     silicoaluminophosphate zeolite prepn org template petroleum refining
    catalyst
     Petroleum refining catalysts
ΙT
        (synthesis of CHA framework type-zeolites as petroleum refining
        catalysts)
ΙT
    Aluminophosphate zeolites
    Silicoaluminophosphate zeolites
        (synthesis of CHA framework type-zeolites as petroleum refining
        catalysts)
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(colloidal, Cabosil; synthesis of CHA framework type-zeolites as petroleum refining catalysts)

IT56-34-8, Tetraethylammonium chloride 71-91-0, Tetraethylammonium 77-98-5, Tetraethylammonium hydroxide 108-00-9, N, N-Dimethylethylenediamine 108-01-0, N, N-Dimethylethanolamine 108-16-7, 1-Dimethylamino-2-propanol : 598-56-1 665-46-3, Tetraethylammonium fluoride 926-63-6, Dimethylpropylamine 1862-07-3, 1-Hexanol, 6-(dimethylamino)-1938-58-5, 1,6-Hexanediamine, N,N-dimethyl 3179-63-3, 1-Propanol, 3529-10-0, 1,4-Butanediamine, N,N-dimethyl 3-(dimethylamino)-5277-11-2, Dimethylheptylamine 4385-04-0 13330-96-6, 1-Butanol, 4-(dimethylamino) - 22078-09-7, 1,7-Heptanediamine, N,N-dimethyl 26153-88-8, N, N-Dimethylpentylamine 27384-67-4, 1-Heptanol, 7-(dimethylamino)-(template; synthesis of CHA framework type-zeolites as petroleum

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AN 2003:730548 HCAPLUS

DN 139:247842

ED Entered STN: 17 Sep 2003

refining catalysts)

TI Synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manufacture of lower alkenes and alkylamines

IN Cao, Guang; Shah, Matu J.; Strohmaier, Karl G.;
Hall, Richard B.

PA Exxonmobil Chemical Patents Inc., USA

SO U.S., 18 pp. CODEN: USXXAM

DT Patent

LA English

IC ICM B01J029-83 ICS B01J029-85

NCL 585640000; 585639000; 502208000; 502214000; 423305000; 423306000; 423327100; 423-DIG.30

CC 51-11 (Fossil Fuels, Derivatives, and Related Products) Section cross-reference(s): 35, 45, 49

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PATENT NO. KIND DATE APPLICATION NO. DATE

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CLASS
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                         PATENT FAMILY CLASSIFICATION CODES
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                        B01J029-83
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                 NCL
                        585640000; 585639000; 502208000; 502214000;
                        423305000; 423306000; 423327100; 423-DIG.30
OS
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AB
     Aluminophosphate zeolites and silicoaluminophosphate zeolites with
     CHA-type framework are prepd. from an alumina source, a compd.
     contg. two or more fluorine substituents that is capable of
     providing fluoride ions, an org. template, and, optionally, a source
     of silica, followed by induced crystn. The template has a general
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structure R1R2N-R3, in which R1 and R2 are C1-5-alkyl; and R3 = C1-12-alkyl, C1-8-cycloalkyl, C1-12-hydroxyalkyl, and C1-12-aminoalkyl. Specific templates include tetraethylammonium salts (selected as OH-, PO43-, F-, Br-, C1-, and OAc-) as well as dipropylamine, isopropylamine, cyclohexylamine, morpholine, methylbutylamine, diethanolamine, or triethylamine. Fluorinated compds. include Et4N.PF6, NaHF2, NH4PF6, H2SiF6, (NH4)2SiF6, NH4HF2, NaPF6, A1F3, (NH4)3A1F6, (NH4)2TiF6, (NH4)2ZrF6, (NH4)2GeF6, and (NH4)2SnF6. The zeolites have catalytic activity for conversion of an oxygenated feedstock, esp. MeOH, to lower olefins (esp. ethylene and propylene) or to alkylamines when reacted in the presence of ammonia.

- ST aluminophosphate silicoaluminophosphate zeolite methanol conversion alkene; alkylamine methanol ammonia conversion aluminophosphate SAPO zeolite
- IT Hydrocarbons, preparation.

(C>4, methanol conversion to; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)

- IT Alkenes, preparation
 - (C2-4, manuf. of, from methanol; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)
- IT Petroleum refining catalysts

(conversion, for methanol; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)

IT Functional groups

(dimethylamino-, in templates; in synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)

IT Amines, preparation

(manuf. of, from methanol; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)

- IT Aluminophosphate zeolites
 - Silicoaluminophosphate zeolites

(synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)

IT Quaternary ammonium compounds, uses

(templates; in synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)

IT Amines, uses

(tertiary, templates; in synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)

- IT 67-56-1, Methanol, processes (conversion of; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)
- IT429-07-2, Tetraethylammonium hexafluorophosphate 1333-83-1, Sodium fluoride (NaHF2) 1341-49-7, Ammonium fluoride ((NH4)HF2) 7784-19-2, Ammonium 7784-18-1, Aluminum trifluoride hexafluoroaluminate 16919-19-0, Ammonium hexafluorosilicate 16919-24-7, Ammonium hexafluorostannate 16919-31-6, Ammonium hexafluorozirconate 16940-81-1, Hydrogen hexafluorophosphate 16941-11-0, Ammonium hexafluorophosphate 16961-83-4 16962-40-6, Ammonium hexafluorotitanate 16962-47-3 21324-39-0, Sodium hexafluorophosphate 32287-65-3, Aluminum trifluoride, hydrate (fluoride source; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)
- TT 74-84-0P, Ethane, preparation 74-98-6P, Propane, preparation 106-98-9P, 1-Butene, preparation 107-01-7P, 2-Butene (methanol conversion to; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)
- IT 115-07-1P, Propylene, preparation
 (methanol conversion to; synthesis of aluminophosphate and
 silicoaluminophosphate zeolites as methanol conversion catalysts
 for manuf. of lower alkenes and alkylamines)
- 56-34-8, Tetraethylammonium chloride 71-91-0, Tetraethylammonium 75-31-0, Isopropylamine, uses 77-98-5, Tetraethylammonium hydroxide 108-00-9, N, N-Dimethylethylenediamine 108-01-0, N, N-Dimethylethanolamine 108-91-8, Cyclohexylamine, uses 109-55-7, N, N-Dimethylpropylenediamine 110-68-9, Methylbutylamine 110-91-8, Morpholine, uses 111-42-2, Diethanolamine, uses 121-44-8, Triethylamine, uses 142-84-7, Dipropylamine 598-56-1 665-46-3, Tetraethylammonium fluoride 926-63-6, Dimethylpropylamine 1185-59-7, Tetraethylammonium acetate 1938-58-5 3529-10-0 4385-04-0 5277-11-2, Dimethylheptylamine 76206-78-5, Ethanaminium, N,N,N-triethyl-, phosphate 22078-09-7
 - (3:1) 154976-19-9, Butanol, (dimethylamino) 154976-21-3, Hexanol, (dimethylamino) 597578-32-0 (templates; synthesis of aluminophosphate and silicoaluminophosphate zeolites as methanol conversion catalysts for manuf. of lower alkenes and alkylamines)
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